

ERRATUM

F. FREEMAN and A. YERAMYAN: Kinetics of the permanganate oxidation of phenylnitromethane
Tetrahedron Letters No.46, pp. 4785-4786 (1968)

p. 4784. Table I. The heading for the sixth column should read $2k_2, M^{-1} \text{ sec}^{-1}$, and the values in this column should be multiplied by 0.666.

p. 4785. The rate law should read

$$\frac{-d[\text{MnO}_4^-]}{dt} = 2k_2[\text{MnO}_4^-][\text{C}_6\text{H}_5\text{CH}=\text{NO}_2^-]$$